

Charges of elements list pdf

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Since supporters of then-President Donald Trump swarmed the US Capitol on January 6, 2021 — forcing Congress to go into lockdown and damaging the halls of government — at least 919 people have been arrested and charged with crimes. The FBI is seeking the public's help to identify people who took part in one of the most documented crimes in US history. But since most rioters were allowed to walk free that day, it's taking some time to track them down. This table includes the names, charges, links to court documents, and links to additional stories of all the people charged so far. We're keeping it updated as the Department of Justice releases more names and info. Only 396 federally charged rioters have entered guilty pleas so far. The number of people charged in the insurrection is expected to keep growing as FBI agents pore through video footage, social media posts, phone location data, and tips from the public. Democratic members of Congress pushed for a bipartisan commission to investigate the Capitol riot akin to the one made post-9/11, but Senate Republicans blocked the attempt. The House of Representatives narrowly voted to create a commission on June 30, 2021. Reps. Liz Cheney and Adam Kinzinger are the only Republicans on the panel. House Minority Leader Kevin McCarthy said the Republican Party would create its own investigation into the January 6, 2021, riot. The committee scheduled its first hearing to showcase its findings on June 9, 2022. On January 13, 2022, the Department of Justice unsealed an indictment against the leader of the Oath Keepers, Elmer Stewart Rhodes, and 10 associates, charging them with "sedition conspiracy," the most severe charge leveled against any of the alleged Capitol rioters so far. The first jury trial relating to the Capitol riots resulted in a guilty finding for rioter Guy Fieri, guilty on each of the five counts he faced after fewer than 4 hours of deliberations. US prosecutors said Fieri brought a pistol and flex-cuffs to the Capitol, but didn't enter the building. Updated: 12/30/2021 by Computer Hope An element may refer to any of the following: 1. When referring to Hypertext Markup Language, see our HTML element definition for further information. 2. Element is often used as an abbreviation of Adobe Photoshop Elements. An element is a single part of a larger comp. For example, a computer programming array can contain different elements (index) that can be stored and used up in a loop. For example, in the below PHP code the @names array contains five elements (names) and the @name variable is used to say hello to each of the names. French for example my @names = ("Neil", "Cathy", "Scott", "Tim"); foreach (@names) { print "Hello \$ " . " " . \$name . "\n"; } In the example above, the following output would be printed to the screen: Hello Neil Hello Cathy Hello Scott Hello Tim This element is one way elements could be added and used in a program. A program could also use an external source such as a file or a database to create a list of elements. Programming terms: The halogen elements are located in group 17 or VIIA of the periodic table, which is the second-to-last column of the chart. This is a list of elements that belong to the halogen group and a look at the properties that they share in common. The halogens are the elements in group 17 of the periodic table. This is the next-to-last column of elements on the righthand side of the table. The halogen elements are fluorine, chlorine, bromine, iodine, and astatine. These are the only group of elements that include elements in all of the three main states of matter: gases, liquids, and solids. Depending on who you ask, there are either 5 or 6 halogens. Fluorine, chlorine, bromine, iodine, and astatine definitely are halogens. Element 117, tennessine, might have some properties in common with the other elements. Even though it is in the same column or group of the periodic table with the other halogens, most scientists believe element 117 behaves more like a metalloid. So little of it has been produced, it's a matter of prediction, not empirical data. These elements share some common properties that distinguish them from other elements on the periodic table. They are highly reactive nonmetals. Atoms belonging to the halogen group have 7 electrons in their outermost (valence) shell. These atoms need one more electron in order to have a stable octet. The usual oxidation state of a halogen atom is -1. Halogens are highly electronegative, with high electron affinities. The melting and boiling points of the halogens increase as you increase atomic number (as you move down the periodic table). The elements change their state of matter at room temperature and pressure as you increase atomic number. Fluorine and chlorine are gases. Bromine is a liquid element. Iodine is a solid. Scientists predict tennessine is a solid. Chlorine is atomic number 17 with element symbol Cl. Under ordinary conditions, chlorine is a yellowish-green gas. Bromine is element 35 with symbol Br. It is a liquid at room temperature and pressure. Iodine is element 53 with symbol I. It is a solid under ordinary conditions. Astatine is atomic number 85 with symbol At. It is the rarest naturally-occurring element in the Earth's crust. Astatine is a radioactive element with no stable isotopes. Tennessine is element 117 with symbol Ts. It is a synthetic radioactive element. The lighter halogens occur in living organisms. These are fluorine, chlorine, bromine, and iodine. Of these, chlorine and iodine are essential for human nutrition, although the other elements might also be required in trace amounts. The halogens are important disinfectants. Chlorine and bromine are used to disinfect water and surfaces. Their high reactivity also makes these elements important components of some types of bleach. Halogens are used in incandescent lamps to make them glow at a higher temperature and with a white color. The halogen elements are important drug components, as they aid drug penetration into tissues. Bonchev, Danail; Kamenska, Verginia (1981). "Predicting the properties of the 113-120 transactinide elements". *The Journal of Physical Chemistry*, 85 (9): 1177-86. doi:10.1021/j150609a021 Emsley, John (2011). *Nature's Building Blocks*. ISBN 978-0-199605637. Greenwood, Norman N.; Earnshaw, Alan (1997). *Chemistry of the Elements* (2nd ed.). Butterworth-Heinemann. ISBN 978-0-08-037941-8. Lide, D. R., ed. (2003). *CRC Handbook of Chemistry and Physics* (84th ed.). Boca Raton, FL: CRC Press. Morss, Lester R.; Edelstein, Norman M.; Fuger, Jean (eds.). *The Chemistry of the Actinide and Transactinide Elements*. Dordrecht, The Netherlands: Springer Science+Business Media. doi:10.1007/978-94-007-0211-0. ISBN 978-94-007-0210-3.

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